## RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College affiliated to University of Calcutta)

## B.A./B.Sc. THIRD SEMESTER EXAMINATION, MARCH 2022 SECOND YEAR [BATCH 2020-23] INDUSTRIAL CHEMISTRY (HONOURS) Paper : VII [CC7]

Date : 07/03/2022 Time : 11 am - 1 pm

## Full Marks : 50

[5×10]

Answer any five questions of the following:

- 1. a) What is coal? How is it formed? What are its types? What are the requirements of a good fuel?
  - b) Calculate the minimum amount of air required for the complete combustion of 50 kgs of coal containing C = 75 %,  $H_2 = 10$  %,  $O_2 = 02$  %, S = 05 %, and the rest nitrogen by weight. [(1+1+1+2)+5]
- 2. a) Describe proximate analysis. Bring out its importance.
  - b) A coal sample on analysis gives C = 75%,  $H_2 = 6\%$ ,  $O_2 = 3.5\%$ , S = 03% and the rest ash. Calculate the Gross and Net calorific values of the fuel. Latent heat of steam is = 587 cal/gm. (5+5)
- 3. a) What is a furnace?
  - b) How many types of furnaces are there?
  - c) What are the parameters to be considered in the design of an efficient furnace?
  - d) What are the major losses in a furnace?
  - e) What are the disadvantages of excess air in a furnace?
- 4. a) What is water gas?
  - b) How is it manufactured?
  - c) Write the uses of water gas.
  - d) Write a brief note on CNG
- 5. a) Define :
  - i) Gross calorific value
  - ii) Net calorific value
  - b) A coal sample with 93% carbon, 5% of Hydrogen and 2% Ash is subjected to combustion in a bomb calorimeter. Calculate GCV and NCV Given that. Mass of the coal sample = 0.95g Mass of water in copper calorimeter = 2000g.

Water equivalent weight of calorimeter = 700g.

Initial temperature  $(t_1) = 30^0 c$ 

Final temperature  $(t_2) = 31.1^{\circ}c$ 

[(2.5+2.5)+5]

(2+2+2+2+2)

(2+2+2+4)

- 6. a) How is producer gas manufactured? What are the uses of producer gas?
  - b) What is flue gas? Discuss Dulong's formula.
  - c) Write the importance of flue gas analysis. [(2+2)+(1+2)+3]
- 7. a) The percentage composition of a sample of coal by weight was found to be C = 72%,

H=6.5% , O=13% , N=2.5% , S=1.8%

- i) Calculate the minimum weight of air for complete combustion.
- ii) Calculate the weight of dry products formed if 50% excess air is used.
- b) Write a Short note on LPG.
- 8. a) What is calorific value? Mention its units. What is the difference between ultimate analysis and proximate analysis of coal?

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b) What is metallurgical coke? Describe Otto Hoffmann by product coke oven method. [(2+2)+(1+5)]

[(1+1+3)+5]

(2)